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[UNIT 2 \(PART 1\) PRACTICE TEST](#)

Honour Chemistry Practice Test: Unit 2 (Part 1): Matter as Solutions and Gases 1 atm = 760 mm Hg = 760 torr = 101.325 kPa TK = TC + 273.15 STP = 22.4 L mol @ 0.00C & 101.325 kPa (1 atm) SATP = 24.8 L mol @ 25.0C & 100. kPa L atm L kPa m m nsolute R = 0.0821 or 8.314 D= n= C= K mol K mol V M Vsolution P1V1 P2V2 r1 M2 n1 P1 C1V1 = C2V2 PV = nRT = = ?1 = = n1T1 n2T2 r2 M1 nTotal PTotal PTotal = P1 +...

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